

S/N 10/554,708

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NEWS	2	JAN 02	STN pricing information for 2008 now available
NEWS	3	JAN 16	CAS patent coverage enhanced to include exemplified prophetic substances
NEWS	4	JAN 28	USPATFULL, USPAT2, and USPATOLD enhanced with new custom IPC display formats
NEWS	5	JAN 28	MARPAT searching enhanced
NEWS	6	JAN 28	USGENE now provides USPTO sequence data within 3 days of publication
NEWS	7	JAN 28	TOXCENTER enhanced with reloaded MEDLINE segment
NEWS	8	JAN 28	MEDLINE and LMEMLINE reloaded with enhancements
NEWS	9	FEB 08	STN Express, Version 8.3, now available
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NEWS	11	FEB 25	IFIREF reloaded with enhancements
NEWS	12	FEB 25	IMSPRODUCT reloaded with enhancements
NEWS	13	FEB 29	WPINDEX/WPIDS/WPIX enhanced with ECLA and current U.S. National Patent Classification
NEWS	14	MAR 31	IFICDB, IFIPAT, and IFIUDB enhanced with new custom IPC display formats
NEWS	15	MAR 31	CAS REGISTRY enhanced with additional experimental spectra
NEWS	16	MAR 31	CA/CAPLUS and CASREACT patent number format for U.S. applications updated
NEWS	17	MAR 31	LPCI now available as a replacement to LDPCI
NEWS	18	MAR 31	EMBASE, EMBAL, and LEMBASE reloaded with enhancements
NEWS	19	APR 04	STN AnaVist, Version 1, to be discontinued
NEWS	20	APR 15	WPIDS, WPINDEX, and WPIX enhanced with new predefined hit display formats
NEWS	21	APR 28	EMBASE Controlled Term thesaurus enhanced
NEWS	22	APR 28	IMSRESEARCH reloaded with enhancements
NEWS	23	MAY 30	INPAFAMDB now available on STN for patent family searching
NEWS	24	MAY 30	DGENE, PCTGEN, and USGENE enhanced with new homology sequence search option
NEWS	25	JUN 06	EPFULL enhanced with 260,000 English abstracts
NEWS	26	JUN 06	KOREAPAT updated with 41,000 documents
NEWS	EXPRESS	FEBRUARY 08	CURRENT WINDOWS VERSION IS V8.3, AND CURRENT DISCOVER FILE IS DATED 20 FEBRUARY 2008
NEWS	HOURS		STN Operating Hours Plus Help Desk Availability
NEWS	LOGIN		Welcome Banner and News Items
NEWS	IPC8		For general information regarding STN implementation of IPC 8

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* * * * * STN Columbus * * * * *

FILE 'HOME' ENTERED AT 23:16:40 ON 08 JUN 2008

=> set abbr on perm
SET COMMAND COMPLETED

=> set plurals on perm
SET COMMAND COMPLETED

=> file uspatall caplus japio		
COST IN U.S. DOLLARS	SINCE FILE	TOTAL
	ENTRY	SESSION
FULL ESTIMATED COST	0.63	0.63

FILE 'USPATFULL' ENTERED AT 23:18:08 ON 08 JUN 2008
CA INDEXING COPYRIGHT (C) 2008 AMERICAN CHEMICAL SOCIETY (ACS)

FILE 'USPATOLD' ENTERED AT 23:18:08 ON 08 JUN 2008
CA INDEXING COPYRIGHT (C) 2008 AMERICAN CHEMICAL SOCIETY (ACS)

FILE 'USPAT2' ENTERED AT 23:18:08 ON 08 JUN 2008
CA INDEXING COPYRIGHT (C) 2008 AMERICAN CHEMICAL SOCIETY (ACS)

FILE 'CAPLUS' ENTERED AT 23:18:08 ON 08 JUN 2008
USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.
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FILE 'JAPIO' ENTERED AT 23:18:08 ON 08 JUN 2008
COPYRIGHT (C) 2008 Japanese Patent Office (JPO)- JAPIO

=> s (imidat? or imidiz? or imidis?) and supercritical?
L1 159 (IMIDAT? OR IMIDIZ? OR IMIDIS?) AND SUPERCRITICAL?

=> s sma (s)(polymer? or copolymer?)
L2 4630 SMA (S) (POLYMER? OR COPOLYMER?)

=> s l1 and l2
L3 0 L1 AND L2

=> s (styren? or vinyl(2a)aromatic)(3a)copoly? or poly(1w)styren?(1w)co?
1 FILES SEARCHED...
TERM 'CO?' EXCEEDED TRUNCATION LIMITS - SEARCH ENDED
You have entered a truncated stem which occurs in too many terms.
Make the stem longer and try again. For example, if your original term was 'degr?' to search for variations and the abbreviation for 'degradation', you could replace it with the expression '(degrdn OR degrad?)'. If your search term was numeric, e.g., 'C>5', reduce the size of the range.

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=> s (styren? or vinyl(2a)aromatic)(3a)copoly? or poly(1w)styren?(3w)maleic anhydride

L4 279533 (STYREN? OR VINYL(2A) AROMATIC)(3A) COPOLY? OR POLY(1W) STYREN?(3W) MALEIC ANHYDRIDE

=> s l1 and l4

L5 11 L1 AND L4

=> d l5 1-11 ibib abs

L5 ANSWER 1 OF 11 USPATFULL on STN

ACCESSION NUMBER: 2008:118581 USPATFULL

TITLE: Ink-jet ink and cured film obtained from same

INVENTOR(S): Satou, Hiroyuki, Ichihara-shi, JAPAN

Deyama, Yoshihiro, Ichihara-shi, JAPAN

Yamahiro, Mikio, Ichihara-shi, JAPAN

PATENT ASSIGNEE(S): CHISSO Corporation (non-U.S. corporation)

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 20080103280	A1	20080501
APPLICATION INFO.:	US 2007-976689	A1	20071026 (11)

	NUMBER	DATE
PRIORITY INFORMATION:	JP 2006-290740	20061026
DOCUMENT TYPE:	Utility	
FILE SEGMENT:	APPLICATION	
LEGAL REPRESENTATIVE:	HOGAN & HARTSON LLP, IP GROUP, COLUMBIA SQUARE, 555 THIRTEENTH STREET, N.W., WASHINGTON, DC, 20004, US	
NUMBER OF CLAIMS:	33	
EXEMPLARY CLAIM:	1	
LINE COUNT:	2088	

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB The invention provides an ink-jet ink including a fluorine-containing compound (C) in the form of fluorosilsesquioxane having an organic group having 1 to 100 carbon atoms.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

L5 ANSWER 2 OF 11 USPATFULL on STN

ACCESSION NUMBER: 2007:140924 USPATFULL

TITLE: Continuous process for the production of combinatorial libraries of modified materials

INVENTOR(S): Nelson, James M., Roseville, MN, UNITED STATES

Marx, Ryan E., Cottage Grove, MN, UNITED STATES

Cernohous, Jeffrey J., Hudson, WI, UNITED STATES

McNerney, James R., Inver Grove Heights, MN, UNITED STATES

PATENT ASSIGNEE(S): 3M Innovative Properties Company (U.S. corporation)

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 20070122915	A1	20070531
APPLICATION INFO.:	US 2006-646856	A1	20061228 (11)
RELATED APPLN. INFO.:	Continuation of Ser. No. US 2002-211219, filed on 2 Aug 2002, GRANTED, Pat. No. US 7157283		
DOCUMENT TYPE:	Utility		
FILE SEGMENT:	APPLICATION		
LEGAL REPRESENTATIVE:	3M INNOVATIVE PROPERTIES COMPANY, PO BOX 33427, ST.		

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PAUL, MN, 55133-3427, US
NUMBER OF CLAIMS: 20
EXEMPLARY CLAIM: 1
LINE COUNT: 1325

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB A system is provided wherein a devolatilizing reactor is used to make combinatorial libraries of materials. Examples of suitable reactors include continuous high viscosity devolatilizers and continuous devolatilizing kneaders.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

L5 ANSWER 3 OF 11 USPATFULL on STN

ACCESSION NUMBER: 2007:107637 USPATFULL

TITLE: Method for preparing styrene and maleimide copolymer using super critical fluid

INVENTOR(S): Kim, Dong Ryul, Daejeon, KOREA, REPUBLIC OF
Park, Sang Hyun, Daejeon, KOREA, REPUBLIC OF
Lee, Hee Hyun, Daejeon, KOREA, REPUBLIC OF

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 20070093610	A1	20070426
APPLICATION INFO.:	US 2004-554708	A1	20040512 (10)
	WO 2004-KR1090		20040512
			20051027 PCT 371 date

DOCUMENT TYPE: Utility

FILE SEGMENT: APPLICATION

LEGAL REPRESENTATIVE: Mckenna Long & Aldridge LLP, 1900 K Street, NW,
Washington, DC, 20006, US

NUMBER OF CLAIMS: 29

EXEMPLARY CLAIM: 1

LINE COUNT: 593

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB Disclosed is a method for preparing copolymers of styrene and maleimide by imidization-extrusion of copolymers of styrene and maleic anhydride using a supercritical fluid. The method can be performed at a lower temperature compared to a conventional method and can remove residual amines and by-products with ease, and thus can provide copolymers of styrene and maleimide having excellent optical properties, thermal properties and mechanical properties.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

L5 ANSWER 4 OF 11 USPATFULL on STN

ACCESSION NUMBER: 2004:231455 USPATFULL

TITLE: Process of manufacturing optical waveguide and connection structure of optical devices

INVENTOR(S): Oe, Kunishige, Kyoto, JAPAN
Yamashita, Kenichi, Kyoto, JAPAN
Mune, Kazunori, Osaka, JAPAN
Mochizuki, Amane, Osaka, JAPAN
Naitou, Ryuusuke, Osaka, JAPAN

PATENT ASSIGNEE(S): NITTO DENKO CORPORATION (U.S. individual)

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 20040178522	A1	20040916
APPLICATION INFO.:	US 2004-760389	A1	20040121 (10)

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	NUMBER	DATE
PRIORITY INFORMATION:	JP 2003-15533	20030124
DOCUMENT TYPE:	Utility	
FILE SEGMENT:	APPLICATION	
LEGAL REPRESENTATIVE:	SUGHRUE MION, PLLC, 2100 PENNSYLVANIA AVENUE, N.W., SUITE 800, WASHINGTON, DC, 20037	
NUMBER OF CLAIMS:	12	
EXEMPLARY CLAIM:	1	
NUMBER OF DRAWINGS:	5 Drawing Page(s)	
LINE COUNT:	928	

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB The invention provides a process of manufacturing an optical waveguide for optically connecting a plurality of optical devices, comprising the steps of: disposing a resin composition between two or more optical devices, the resin composition comprising a resin and a 1,4-dihydropyridine derivative, forming an optical path through the resin composition between the optical devices by light having a wavelength capable of inducing a structural change in the 1,4-dihydropyridine derivative, and removing the 1,4-dihydropyridine derivative from the resulting resin composition. Also disclosed is a connection structure obtained by the process.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

L5 ANSWER 5 OF 11 USPATFULL on STN

ACCESSION NUMBER: 2004:31988 USPATFULL

TITLE: Process to modify polymeric materials and resulting compositions

INVENTOR(S): Nelson, James M., Roseville, MN, UNITED STATES
Marx, Ryan E., Cottage Grove, MN, UNITED STATES
Cernohous, Jeffrey J., Hudson, WI, UNITED STATES
McNerney, James R., Inver Grove Heights, MN, UNITED STATES
Jones, Todd D., St. Paul, MN, UNITED STATES
Hanley, Kenneth J., Eagan, MN, UNITED STATES

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 20040024130	A1	20040205
APPLICATION INFO.:	US 2002-211415	A1	20020802 (10)
DOCUMENT TYPE:	Utility		
FILE SEGMENT:	APPLICATION		
LEGAL REPRESENTATIVE:	3M INNOVATIVE PROPERTIES COMPANY, PO BOX 33427, ST. PAUL, MN, 55133-3427		
NUMBER OF CLAIMS:	39		
EXEMPLARY CLAIM:	1		
LINE COUNT:	1799		

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB Disclosed is a method for modifying a polymer by carrying out a thermally-induced reaction in a mixing apparatus having a high shear environment and devolatilization capabilities. Also disclosed are the resulting materials.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

L5 ANSWER 6 OF 11 USPATFULL on STN

ACCESSION NUMBER: 2004:31259 USPATFULL

TITLE: Continuous process for the production of combinatorial

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libraries of modified materials
INVENTOR(S): Nelson, James M., Roseville, MN, UNITED STATES
Marx, Ryan E., Cottage Grove, MN, UNITED STATES
Cernohous, Jeffrey J., Hudson, WI, UNITED STATES
McNerney, James R., Inver Grove Heights, MN, UNITED STATES

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 20040023398	A1	20040205
	US 7157283	B2	20070102
APPLICATION INFO.:	US 2002-211219	A1	20020802 (10)
DOCUMENT TYPE:	Utility		
FILE SEGMENT:	APPLICATION		
LEGAL REPRESENTATIVE:	3M INNOVATIVE PROPERTIES COMPANY, PO BOX 33427, ST. PAUL, MN, 55133-3427		
NUMBER OF CLAIMS:	27		
EXEMPLARY CLAIM:	1		
LINE COUNT:	1342		

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB A system is provided wherein a devolatilizing reactor is used to make combinatorial libraries of materials. Examples of suitable reactors include continuous high viscosity devolatilizers and continuous devolatilizing kneaders.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

L5 ANSWER 7 OF 11 USPATFULL on STN

ACCESSION NUMBER: 2002:337815 USPATFULL

TITLE: Golf ball compositions including microcellular materials and methods for making same

INVENTOR(S): Harris, Kevin M., New Bedford, MA, UNITED STATES
Rajagopalan, Murali, South Dartmouth, MA, UNITED STATES
Cavallaro, Christopher, Lakeville, MA, UNITED STATES

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 20020193179	A1	20021219
	US 6676866	B2	20040113
APPLICATION INFO.:	US 2002-142168	A1	20020510 (10)
RELATED APPLN. INFO.:	Division of Ser. No. US 2000-565108, filed on 4 May 2000, GRANTED, Pat. No. US 6386992		
DOCUMENT TYPE:	Utility		
FILE SEGMENT:	APPLICATION		
LEGAL REPRESENTATIVE:	SWIDLER BERLIN SHEREFF FRIEDMAN, LLP, 3000 K STREET, NW, BOX IP, WASHINGTON, DC, 20007		
NUMBER OF CLAIMS:	24		
EXEMPLARY CLAIM:	1		
NUMBER OF DRAWINGS:	1 Drawing Page(s)		
LINE COUNT:	1034		

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB This invention is directed to golf balls including one or more foamed, microcellular materials. The invention also encompasses methods of controlling or adjusting one or more material properties or the weight distribution of a golf ball, and methods of forming golf balls including such microcellular materials.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

L5 ANSWER 8 OF 11 USPATFULL on STN

S/N 10/554,708

ACCESSION NUMBER: 2002:108265 USPATFULL
TITLE: Golf ball compositions including microcellular materials and methods for making same
INVENTOR(S): Harris, Kevin M., New Bedford, MA, United States
Rajagopalan, Murali, South Dartmouth, MA, United States
Cavallaro, Christopher, Lakeville, MA, United States
PATENT ASSIGNEE(S): Acushnet Company, Fairhaven, MA, United States (U.S. corporation)

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 6386992	B1	20020514
APPLICATION INFO.:	US 2000-565108		20000504 (9)
DOCUMENT TYPE:	Utility		
FILE SEGMENT:	GRANTED		
PRIMARY EXAMINER:	Sewell, Paul T.		
ASSISTANT EXAMINER:	Hunter, Jr., Alvin A.		
LEGAL REPRESENTATIVE:	Swidler Berlin Shereff Friedman, LLP		
NUMBER OF CLAIMS:	14		
EXEMPLARY CLAIM:	1		
NUMBER OF DRAWINGS:	3 Drawing Figure(s); 1 Drawing Page(s)		
LINE COUNT:	963		

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB This invention is directed to golf balls including one or more foamed, microcellular materials. The invention also encompasses methods of controlling or adjusting one or more material properties or the weight distribution of a golf ball, and methods of forming golf balls including such microcellular materials.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

L5 ANSWER 9 OF 11 USPAT2 on STN

ACCESSION NUMBER: 2004:31259 USPAT2
TITLE: Continuous process for the production of combinatorial libraries of modified materials
INVENTOR(S): Nelson, James M., Roseville, MN, UNITED STATES
Marx, Ryan E., Cottage Grove, MN, UNITED STATES
Cernohous, Jeffrey J., Hudson, WI, UNITED STATES
McNerney, James R., Inver Grove Heights, MN, UNITED STATES
PATENT ASSIGNEE(S): 3M Innovative Properties Company, St. Paul, MN, UNITED STATES (U.S. corporation)

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 7157283	B2	20070102
APPLICATION INFO.:	US 2002-211219		20020802 (10)
DOCUMENT TYPE:	Utility		
FILE SEGMENT:	GRANTED		
PRIMARY EXAMINER:	Warden, Jill		
ASSISTANT EXAMINER:	Levkovich, Natalia		
LEGAL REPRESENTATIVE:	Edman, Sean J.		
NUMBER OF CLAIMS:	25		
EXEMPLARY CLAIM:	1		
LINE COUNT:	1396		

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB A system is provided wherein a devolatilizing reactor is used to make combinatorial libraries of materials. Examples of suitable reactors include continuous high viscosity devolatilizers and continuous devolatilizing kneaders.

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CAS INDEXING IS AVAILABLE FOR THIS PATENT.

L5 ANSWER 10 OF 11 USPAT2 on STN
ACCESSION NUMBER: 2002:337815 USPAT2
TITLE: Golf ball compositions including microcellular materials and methods for making same
INVENTOR(S): Harris, Kevin M., New Bedford, MA, United States
Rajagopalan, Murali, South Dartmouth, MA, United States
Cavallaro, Christopher, Lakeville, MA, United States
PATENT ASSIGNEE(S): Acushnet Company, Fairhaven, MA, United States (U.S. corporation)

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 6676866	B2	20040113
APPLICATION INFO.:	US 2002-142168		20020510 (10)
RELATED APPLN. INFO.:	Division of Ser. No. US 2000-565108, filed on 4 May 2000, now patented, Pat. No. US 6386992		
DOCUMENT TYPE:	Utility		
FILE SEGMENT:	GRANTED		
PRIMARY EXAMINER:	Kuhns, Allan R.		
LEGAL REPRESENTATIVE:	Swidler Berlin Shereff Friedman, LLP		
NUMBER OF CLAIMS:	16		
EXEMPLARY CLAIM:	1		
NUMBER OF DRAWINGS:	3 Drawing Figure(s); 1 Drawing Page(s)		
LINE COUNT:	973		

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB This invention is directed to golf balls including one or more foamed, microcellular materials. The invention also encompasses methods of controlling or adjusting one or more material properties or the weight distribution of a golf ball, and methods of forming golf balls including such microcellular materials.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

L5 ANSWER 11 OF 11 CAPLUS COPYRIGHT 2008 ACS on STN
ACCESSION NUMBER: 2004:1016086 CAPLUS
DOCUMENT NUMBER: 141:424600
TITLE: Method for preparing styrene and maleimide copolymer using supercritical fluids
INVENTOR(S): Kim, Dong-Ryul; Park, Sang-Hyun; Lee, Hee-Hyun
PATENT ASSIGNEE(S): LG Chem, Ltd., S. Korea
SOURCE: PCT Int. Appl., 22 pp.
CODEN: PIXXD2
DOCUMENT TYPE: Patent
LANGUAGE: English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2004101636	A2	20041125	WO 2004-KR1090	20040512
WO 2004101636	A3	20050217		
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				

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RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM,
AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK,
EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE,
SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE,
SN, TD, TG

KR 2004098896	A	20041126	KR 2003-31214	20030516
EP 1625170	A2	20060215	EP 2004-732516	20040512
EP 1625170	B1	20070704		

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
IE, SI, FI, RO, CY, TR, BG, CZ, EE, HU, PL, SK

CN 1780864	A	20060531	CN 2004-80011161	20040512
JP 2006522171	T	20060928	JP 2006-500694	20040512
AT 366265	T	20070715	AT 2004-732516	20040512
TW 263649	B	20061011	TW 2004-93113608	20040514
US 20070093610	A1	20070426	US 2005-554708	20051027

PRIORITY APPLN. INFO.:

KR 2003-31214	A	20030516
WO 2004-KR1090	W	20040512

AB Disclosed is a method for preparing copolymers of styrene and maleimide by imidization-extrusion of copolymers of styrene and maleic anhydride using a supercrit. fluid. The method can be performed at a lower temperature compared to a conventional method and can remove residual amines and byproducts with ease, and thus can provide copolymers of styrene and maleimide having excellent optical properties, thermal properties and mech. properties.

=> d 15 5 ibib hit

L5 ANSWER 5 OF 11 USPATFULL on STN

ACCESSION NUMBER: 2004:31988 USPATFULL

TITLE: Process to modify polymeric materials and resulting compositions

INVENTOR(S): Nelson, James M., Roseville, MN, UNITED STATES
Marx, Ryan E., Cottaga Grove, MN, UNITED STATES
Cernohous, Jeffrey J., Hudson, WI, UNITED STATES
McNerney, James R., Inver Grove Heights, MN, UNITED STATES
Jones, Todd D., St. Paul, MN, UNITED STATES
Hanley, Kenneth J., Eagan, MN, UNITED STATES

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 20040024130	A1	20040205
APPLICATION INFO.:	US 2002-211415	A1	20020802 (10)
DOCUMENT TYPE:	Utility		
FILE SEGMENT:	APPLICATION		
LEGAL REPRESENTATIVE:	3M INNOVATIVE PROPERTIES COMPANY, PO BOX 33427, ST. PAUL, MN, 55133-3427		
NUMBER OF CLAIMS:	39		
EXEMPLARY CLAIM:	1		
LINE COUNT:	1799		

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

DETD [0046] One aspect of the present invention employs thermally-induced reactions to modify polymeric materials. Many types of thermally-induced reactions are suitable for the present invention. One suitable type of reaction is a rearrangement reaction in which the substituents or moieties of a molecule are rearranged to form a new molecule, i.e., the bonding site of a substituent or moiety moves from one atom to another in the same molecule. Another suitable type of reaction is an

elimination reaction in which one or more substituents is removed from a molecule. Specific types of reactions that can be carried out include, but are not limited to, pyrolysis reactions, acid-catalyzed reactions, deprotection reactions, condensation reactions, hydrolysis reactions, imidization reactions, base-catalyzed reactions, and deesterification, e.g., deacetylation. In a pyrolysis reaction, a complex molecule is broken into simpler units by the use of heat. In an acid-catalyzed reaction, acid is used to drive or induce the thermal reaction. In a deprotection reaction, a protecting group is removed to expose a reactive functional group. In a condensation reaction, two molecules react to form a new molecule and release a byproduct, which is typically water. In a hydrolysis reaction, water reacts with another molecule (e.g., ester) to form one or more new molecules. In an imidization reaction, anhydrides react with primary amines via an intermediate amic acid functionality to form an imide ring and water. In a base-catalyzed reaction, base is used to drive or induce the thermal reaction. In a deesterification reaction, an ester is converted into a carboxylic acid or an anhydride. In a deacetylation reaction, an ester is converted into an alcohol with removal of an acetyl group. See, for example, Hawker et al., *Macromolecules*, 1998, 31, 1024.

DETD [0066] The starting polymer systems may be synthesized in processes that are carried out in batch, semibatch, continuous stirred tank reactor (CSTR), tubular reactors, stirred tubular reactors, plug flow reactors (PFR), temperature controlled stirred tubular reactors as described in WO 0158962 A1 and co-pending U.S. patent app. Ser. No. 09/824,330, static mixers, continuous loop reactor, extruders, shrouded extruders as described in WO 9740929, and pouched reactors as described in WO 9607522 and WO 9607674. The media in which the polymerizations may take place are bulk, solution, suspension, emulsion, ionic liquids or supercritical fluids, such as supercritical carbon dioxide.

DETD [0075] where represents a bond in a polymerizable or polymer chain; R.sub.f is --C.sub.6F.sub.13, --C.sub.4F.sub.9, or --C.sub.3F.sub.7; R and R.sub.2 are each independently hydrogen or alkyl of 1 to 20 carbon atoms; n is an integer from 2 to 11; and x is an integer of at least 1. An example of a Formula I structure is N-methylperfluorobutanesulfonamide. The fluorinated materials may be end-functionalized at one or both terminus with reactive end groups. If there are two reactive end groups, they may be the same or different. Fluorinated diene, methacrylate and styrenic homo and block copolymers end-functionalized with alcohol(s), thiol(s), and/or amine(s) may be synthesized anionically by the use of suitable anionic initiators which contain protected functional groups that can be removed by post polymerization techniques. Suitable functional initiators are known in the art and are described in, e.g., U.S. Pat. No. 6,197,891, U.S. Pat. No. 6,160,054, U.S. Pat. No. 6,221,991, and U.S. Pat. No. 6,184,338.

DETD [0081] Various reactions may be carried out to produce acetylene-containing polymers. In these reactions, a sulfoxide is pyrolyzed to give the polyacetylene and a sulfenic acid byproduct (RSOH) as shown in Formula II. For example, a benzenesulfenic acid may be eliminated from poly(phenyl vinyl sulfoxide)-containing copolymers to produce polyacetylene-containing copolymers, such as poly(styrene-acetylene) block copolymers. Polyacetylene is typically difficult to work with because it is very insoluble in other materials. However, including it in a block structure allows the final structure to remain soluble.

DETD [0094] Diene, methacrylate and styrenic homo and block copolymers end-functionalized with alcohol(s), thiol(s), and/or amine(s) may be synthesized anionically by the use of suitable anionic initiators which contain protected functional groups that can be removed

by post polymerization techniques. Suitable functional initiators are known in the art and are described in, e.g., U.S. Pat. No. 6,197,891, U.S. Pat. No. 6,160,054, U.S. Pat. No. 6,221,991, and U.S. Pat. No. 6,184,338.

DETD [0106] The polymeric materials produced by acid-catalyzed pyrolysis of methacrylic and acrylic esters are methacrylic/acrylic acid or methacrylic/acrylic anhydride functionalized polymers. These acid- and anhydride-functionalized polymers may participate in further grafting reactions including esterification, amidation, and imidization reactions.

DETD [0108] In the case of amidation or imidization, the acid- or anhydride-functionalized polymeric material is subjected to reaction with amine nucleophiles. Suitable amines that participate in this reaction consist of, but are not limited to, typically primary alkyl, aryl, and alkylaryl-amines. The primary amines formula is RNH_2 wherein R stands for an alkyl group having 1 to 10 carbon atoms, a cycloalkyl group having 5 to 12 carbon atoms, an aralkyl group having 7 to 22 carbon atoms or an aryl group having 6 to 12 carbon atoms.

DETD [0110] Diene, methacrylate and styrenic homo and block copolymers end-functionalized with alcohol(s), thiol(s), and/or amine(s) may be synthesized anionically by the use of suitable anionic initiators, which contain protected functional groups that can be removed by post polymerization techniques. Suitable anionic initiators are known in the art and are described in, e.g., U.S. Pat. No. 6,197,891, U.S. Pat. No. 6,160,054, U.S. Pat. No. 6,221,991, and U.S. Pat. No. 6,184,338.

DETD Batch Synthesis of poly(styrene-acetylene) Block Copolymers via the Thermal Modification of poly(styrene-vinyl phenyl sulfoxide).

DETD [0144] This example illustrates the thermal elimination of benzenesulfenic acid from poly(styrene-vinyl phenyl sulfoxide) (P(S-VPS)) to produce a poly(styrene-acetylene) block copolymer.

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L5 ANSWER 6 OF 11 USPATFULL on STN

ACCESSION NUMBER: 2004:31259 USPATFULL

TITLE: Continuous process for the production of combinatorial libraries of modified materials

INVENTOR(S): Nelson, James M., Roseville, MN, UNITED STATES
Marx, Ryan E., Cottage Grove, MN, UNITED STATES
Cernohous, Jeffrey J., Hudson, WI, UNITED STATES
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	NUMBER	KIND	DATE
PATENT INFORMATION:	US 20040023398	A1	20040205
	US 7157283	B2	20070102
APPLICATION INFO.:	US 2002-211219	A1	20020802 (10)
DOCUMENT TYPE:	Utility		
FILE SEGMENT:	APPLICATION		
LEGAL REPRESENTATIVE:	3M INNOVATIVE PROPERTIES COMPANY, PO BOX 33427, ST. PAUL, MN, 55133-3427		
NUMBER OF CLAIMS:	27		
EXEMPLARY CLAIM:	1		
LINE COUNT:	1342		

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

SUMM [0063] One aspect of the present invention employs thermally-induced reactions to modify polymeric materials. Many types of thermally-induced reactions are suitable for the present invention. One suitable type of reaction is a rearrangement reaction in which the substituents or moieties of a molecule are rearranged to form a new molecule, i.e., the bonding site of a substituent or moiety moves from one atom to another in the same molecule. Another suitable type of reaction is an elimination reaction in which one or more substituents is removed from a molecule. Specific types of reactions that can be carried out include, but are not limited to, pyrolysis reactions, acid-catalyzed reactions, deprotection reactions, condensation reactions, hydrolysis reactions, imidization reactions, base-catalyzed reactions, and deesterification, e.g., deacetylation. In a pyrolysis reaction, a complex molecule is broken into simpler units by the use of heat. In an acid-catalyzed reaction, acid is used to drive or induce the thermal reaction. In a deprotection reaction, a protecting group is removed to expose a reactive functional group. In a condensation reaction, two molecules react to form a new molecule and release a byproduct, which is typically water. In a hydrolysis reaction, water reacts with another molecule (e.g., ester) to form one or more new molecules. In an imidization reaction, anhydrides react with primary amines via an intermediate amic acid functionality to form an imide ring and water. In a base-catalyzed reaction, base is used to drive or induce the thermal reaction. In a deesterification reaction, an ester is converted into a carboxylic acid or an anhydride. In a deacetylation reaction, an ester is converted into an alcohol with removal of an acetyl group. See, for example, Hawker et al., *Macromolecules*, 1998, 31, 1024.

SUMM [0080] The starting polymer systems may be synthesized in processes that are carried out in batch, semibatch, continuous stirred tank reactor (CSTR), tubular reactors, stirred tubular reactors, plug flow reactors (PFR), temperature controlled stirred tubular reactors as described in WO 0158962 A1 and co-pending U.S. patent app. Ser. No. 09/824,330, static mixers, continuous loop reactor, extruders, shrouded extruders as described in WO 9740929, and pouched reactors as described in WO 9607522 and WO 9607674. The media in which the polymerizations may take place are bulk, solution, suspension, emulsion, ionic liquids or supercritical fluids, such as supercritical carbon dioxide.

SUMM [0090] represents a bond in a polymerizable or polymer chain; R.sub.f is --C.sub.6F.sub.13, --C.sub.4F.sub.9, or --C.sub.3F.sub.7; R and R.sub.2 are each independently hydrogen or alkyl of 1 to 20 carbon atoms; n is an integer from 2 to 11; and x is an integer of at least 1. The fluorinated materials may be end-functionalized at one or both terminus with reactive end groups. If there are two reactive end groups, they may be the same or different. Fluorinated diene, methacrylate and styrenic homo and block copolymers end-functionalized with alcohol(s), thiol(s), and/or amine(s) may be synthesized anionically by the use of suitable anionic initiators which contain protected functional groups that can be removed by post polymerization techniques. Suitable functional initiators are known in the art and are described in, e.g., U.S. Pat. No. 6,197,891, U.S. Pat. No. 6,160,054, U.S. Pat. No. 6,221,991, and U.S. Pat. No. 6,184,338.

SUMM [0095] Various reactions may be carried out to produce acetylene-containing polymers. In these reactions, a sulfoxide is pyrolyzed to give the polyacetylene and a sulfenic acid byproduct (RSOH) as shown in Formula II. For example, a benzenesulfenic acid may be eliminated from poly(phenyl vinyl sulfoxide)-containing copolymers to

produce polyacetylene-containing copolymers, such as poly(styrene-acetylene) block copolymers. Polyacetylene is typically difficult to work with because it is very insoluble in other materials. However, including it in a block structure allows the final structure to remain soluble.

SUMM [0108] Diene, methacrylate and styrenic homo and block copolymers end-functionalized with alcohol(s), thiol(s), and/or amine(s) may be synthesized anionically by the use of suitable anionic initiators which contain protected functional groups that can be removed by post polymerization techniques. Suitable functional initiators are known in the art and are described in, e.g., U.S. Pat. No. 6,197,891, U.S. Pat. No. 6,160,054, U.S. Pat. No. 6,221,991, and U.S. Pat. No. 6,184,338.

SUMM [0119] The polymeric materials produced by acid-catalyzed pyrolysis of methacrylic and acrylic esters are methacrylic/acrylic acid or methacrylic/acrylic anhydride functionalized polymers. These acid- and anhydride-functionalized polymers may participate in further grafting reactions including esterification, amidation, and imidization reactions.

SUMM [0121] In the case of amidation or imidization, the acid- or anhydride-functionalized polymeric material is subjected to reaction with amine nucleophiles. Suitable amines that participate in this reaction consist of, but are not limited to, typically primary alkyl, aryl, and alkylaryl-amines. The primary amines formula is RNH_2 wherein R stands for an alkyl group having 1 to 10 carbon atoms, a cycloalkyl group having 5 to 12 carbon atoms, an aralkyl group having 7 to 22 carbon atoms or an aryl group having 6 to 12 carbon atoms.

SUMM [0123] Diene, methacrylate and styrenic homo and block copolymers end-functionalized with alcohol(s), thiol(s), and/or amine(s) may be synthesized anionically by the use of suitable anionic initiators, which contain protected functional groups that can be removed by post polymerization techniques. Suitable anionic initiators are known in the art and are described in, e.g., U.S. Pat. No. 6,197,891, U.S. Pat. No. 6,160,054, U.S. Pat. No. 6,221,991, and U.S. Pat. No. 6,184,338. Objects and advantages of this invention are further illustrated by the following examples. The particular materials and amounts thereof, as well as other conditions and details, recited in these examples should not be used to unduly limit this invention.

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FILE 'USPATFULL, USPATOLD, USPAT2, CAPLUS, JAPIO' ENTERED AT 23:18:08 ON 08 JUN 2008

L1 159 S (IMIDAT? OR IMIDIZ? OR IMIDIS?) AND SUPERCRITICAL?
L2 4630 S SMA (S) (POLYMER? OR COPOLYMER?)
L3 0 S L1 AND L2
L4 279533 S (STYREN? OR VINYL(2A)AROMATIC) (3A)COPOLY? OR POLY(1W)STYREN?
L5 11 S L1 AND L4

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L5 ANSWER 10 OF 11 USPAT2 on STN

ACCESSION NUMBER: 2002:337815 USPAT2

TITLE: Golf ball compositions including microcellular materials and methods for making same

INVENTOR(S): Harris, Kevin M., New Bedford, MA, United States
Rajagopalan, Murali, South Dartmouth, MA, United States
Cavallaro, Christopher, Lakeville, MA, United States

PATENT ASSIGNEE(S): Acushnet Company, Fairhaven, MA, United States (U.S. corporation)

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 6676866	B2	20040113
APPLICATION INFO.:	US 2002-142168		20020510 (10)
RELATED APPLN. INFO.:	Division of Ser. No. US 2000-565108, filed on 4 May 2000, now patented, Pat. No. US 6386992		
DOCUMENT TYPE:	Utility		
FILE SEGMENT:	GRANTED		
PRIMARY EXAMINER:	Kuhns, Allan R.		
LEGAL REPRESENTATIVE:	Swidler Berlin Shereff Friedman, LLP		
NUMBER OF CLAIMS:	16		
EXEMPLARY CLAIM:	1		
NUMBER OF DRAWINGS:	3 Drawing Figure(s); 1 Drawing Page(s)		
LINE COUNT:	973		

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

DETD 1. A supercritical fluid (SCF) of an atmospheric gas is injected into the polymer through a barrel to form a single-phase solution, with the SCF delivery system, screw, and injectors designed to allow for rapid dissolution;

DETD Block polystyrene TPEs suitable for use in this invention include blocks of polystyrene or substituted polystyrene, e.g., poly(α -methyl styrene) or poly(4-methyl styrene) chemically linked or joined to the ends of lower softening point blocks of either an unsaturated or saturated rubber. Unsaturated rubber types typically include butadiene, which can form styrene-butadiene-styrene (hereafter "SBS") block copolymers, or isoprene, which can form styrene-isoprene-styrene (hereafter "SIS") block copolymers, silicone rubber, balata, styrene-butadiene rubber ("SBR"), and the like. Examples of suitable commercially available thermoplastic SBS or SIS copolymers include the KRATON® D series from Shell Corporation of Houston, Tex., which includes KRATON® D2109, D5119 and D5298; VECTOR® from Dexco of Plaquemine, La.; and FINAPRENE® from Fina Oil and Chemical of Plano, Tex.

DETD Any conventional material or method may also be used in preparing the golf ball cover, which is typically disposed over the center or core. For example, as is well known in the art, ionomers, balata, and urethanes are all suitable golf ball cover materials. A variety of less conventional materials may also be used for the cover, e.g., thermoplastics such as ethylene- or propylene-based homopolymers and copolymers. These homopolymers and copolymers may also include functional monomers such as acrylic and methacrylic acid, fully or partially neutralized ionomers and their blends, methyl acrylate, methyl methacrylate homopolymers and copolymers, imidized amino group-containing polymers, polycarbonate, reinforced polycarbonate, reinforced polyamides, polyphenylene oxide, high impact polystyrene, polyether ketone, polysulfone, poly(phenylene sulfide), acrylonitrile-butadiene, acrylic-styrene-terephthalate, poly(ethylene terephthalate), poly(butylene terephthalate), poly(ethylene-vinyl

S/N 10/554,708

alcohol), poly(tetrafluoroethylene), and the like. Any of these polymers or copolymers may be further reinforced by blending with a wide range of density-adjusting fillers, including glass fibers or spheres, or metallic powders. The selection of a suitable cover, and application thereof over the mantle described herein, will be readily determinable by those of ordinary skill in the art, particularly in view of the disclosure herein.

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COST IN U.S. DOLLARS

SINCE FILE

TOTAL

ENTRY

SESSION

FULL ESTIMATED COST

91.21

91.84

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)

SINCE FILE

TOTAL

ENTRY

SESSION

CA SUBSCRIBER PRICE

-0.80

-0.80

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